

# Primrose Creek Stream Analysis

**Solebury, PA**

**2011**

## Primrose Creek Watershed Association

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P.O. Box 49 ▪ Solebury, PA 18963 ▪ [www.primrosecreek.org](http://www.primrosecreek.org) ▪ [primrosecreek@gmail.com](mailto:primrosecreek@gmail.com)

All evaluations were performed by PCWA volunteers utilizing entry level tools. It is not meant to be compared with the accuracy of professional studies. However, we believe with the level of dedication and focus the teams exhibited, this analysis is the start of a database that present general trends and basic understanding of the health of Primrose Creek. Since this effort is a learning process for all involved, techniques and data will evolve.

Primrose Creek was sampled at the following two locations:

East Station Phillips Mill

Approximate Latitude Longitude - 40.38545, -74.96235

West of Solebury School

Approximate Latitude Longitude - 40.37957, -74.99637

The team was comprised of the following members:

East Team

David Harrison

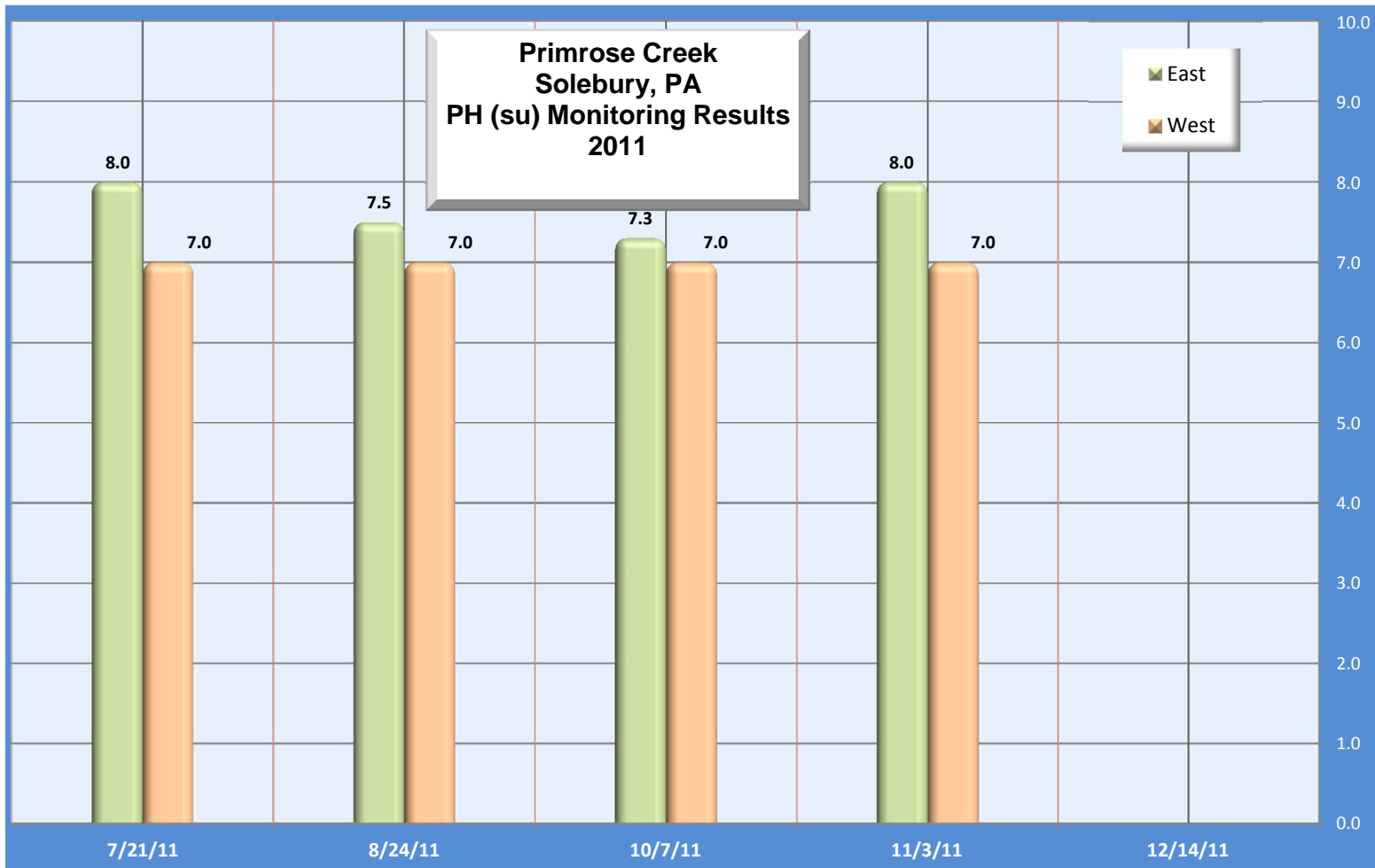
Robert Long

West Team

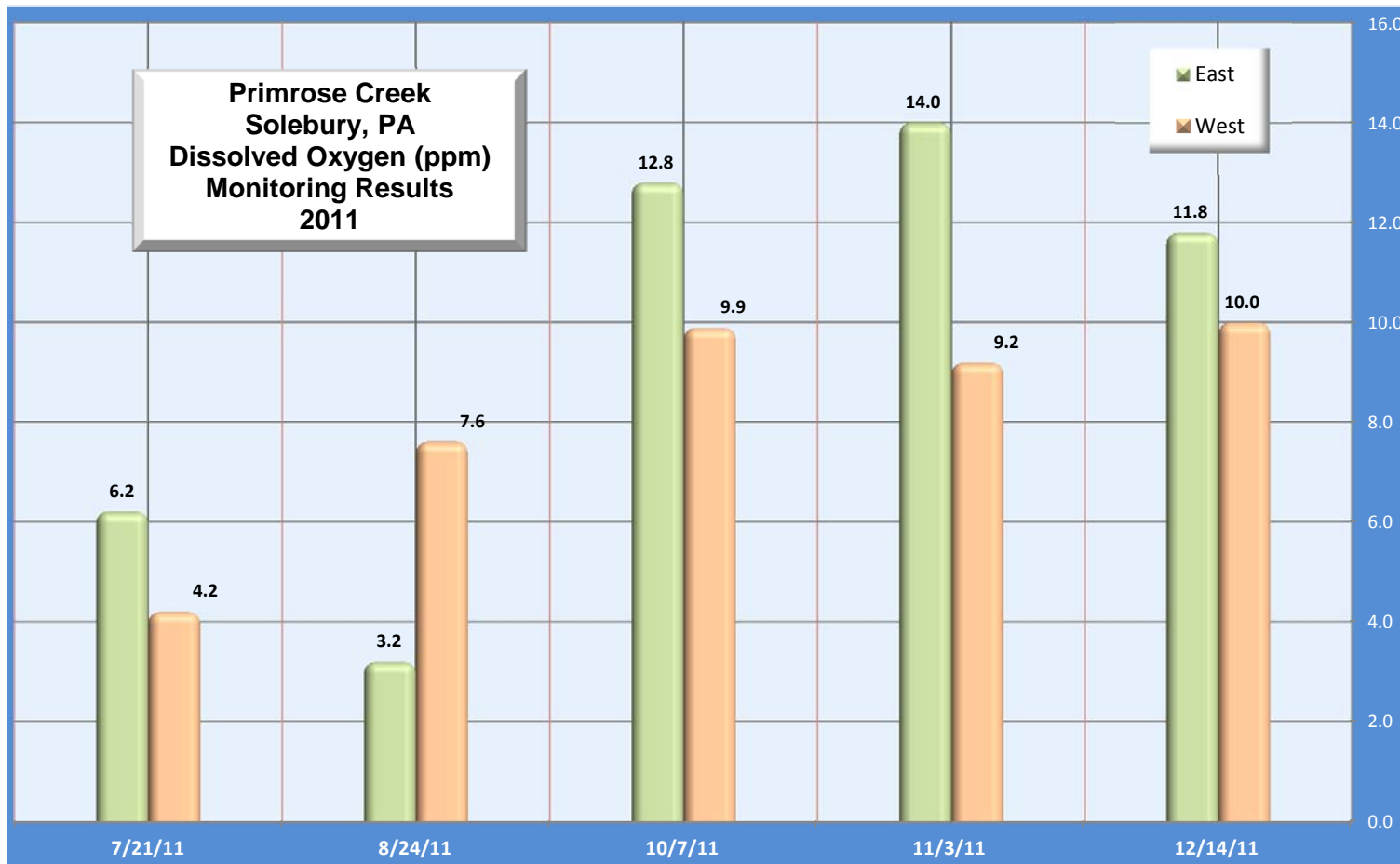
Charles Furst

Carol Cope

## **Graph Trends**



PH is the standard measure of how acidic or alkaline a solution is. It is measured on a scale from 0 – 14 pH of 7 is neutral, pH less than 7 is acidic, and PH greater than 7 is basic. Although each organism has an ideal pH, most aquatic organisms prefer PH of 6.5 – 8.0.

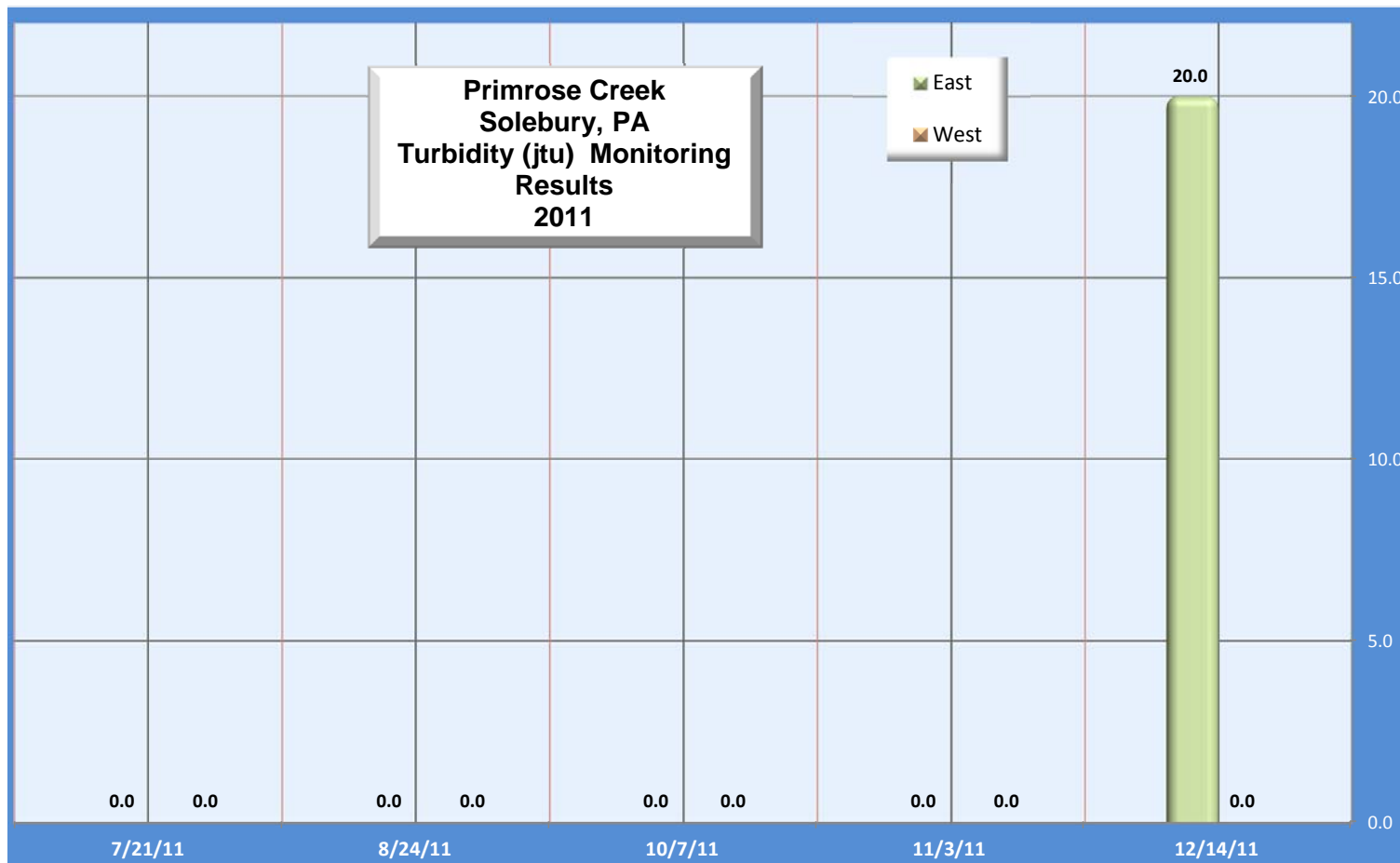


Oxygen is critical to the survival of aquatic plants and animals.

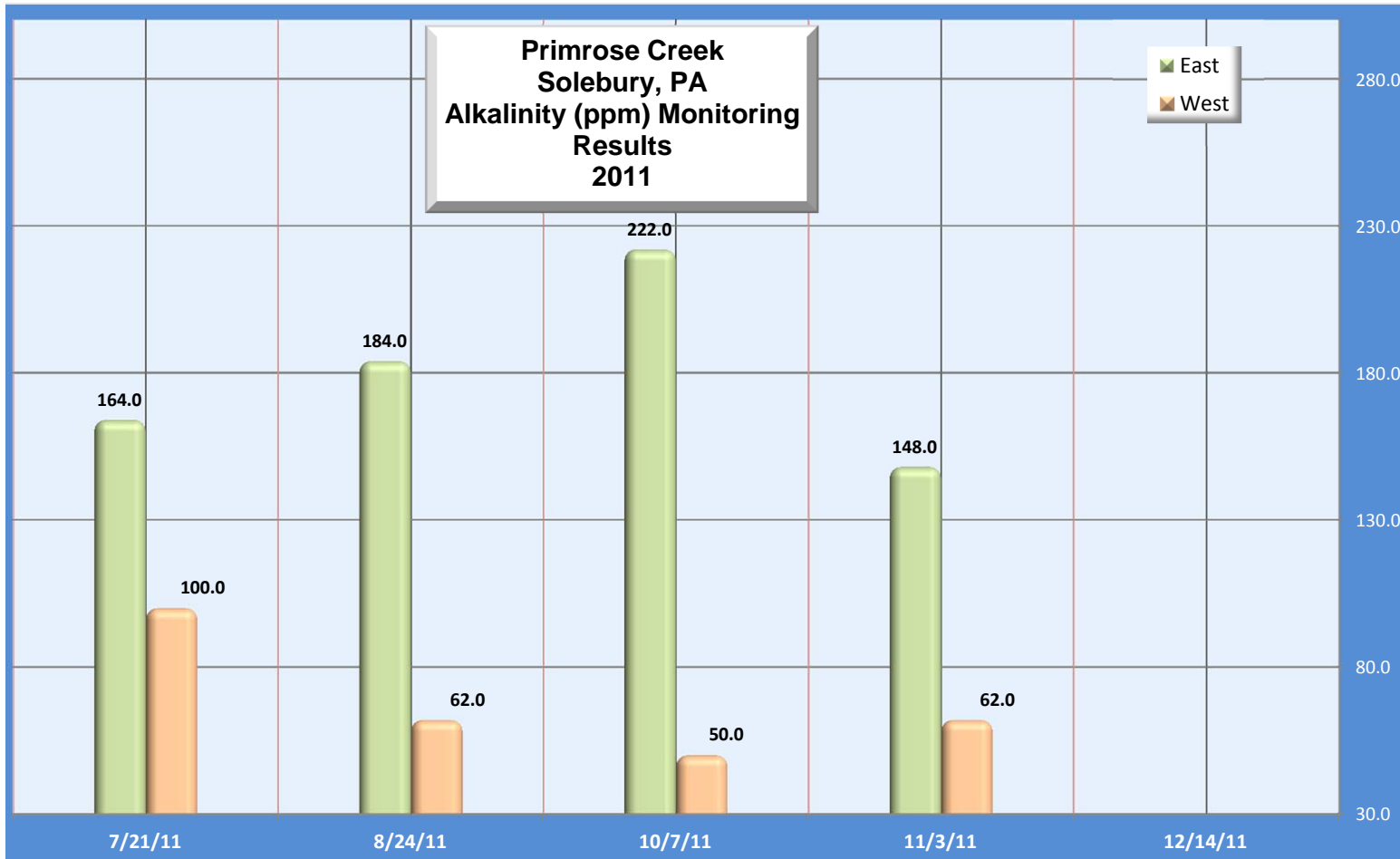
5-6 ppm sufficient for most species

<3 ppm Stressful to most species

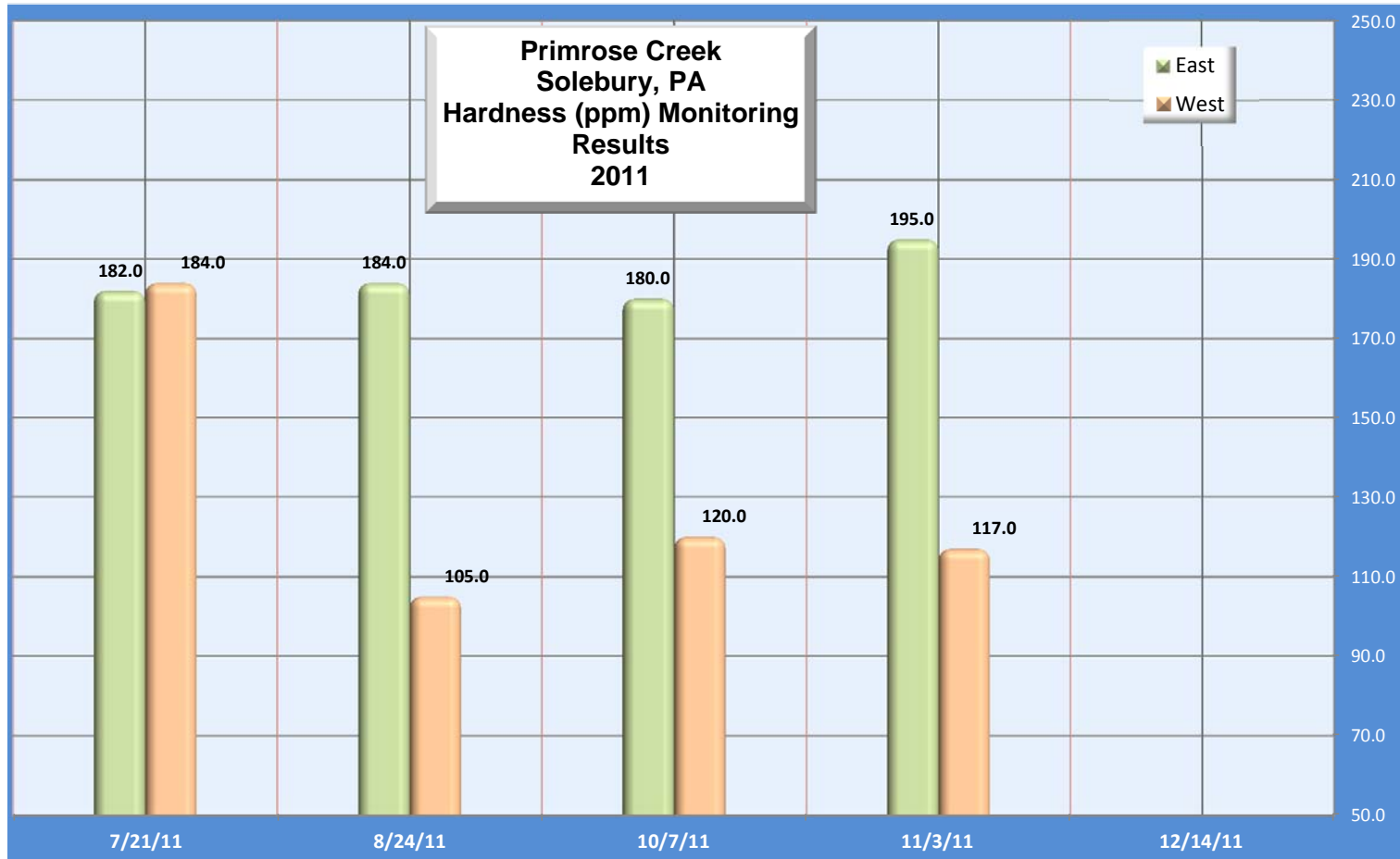
<2 ppm Fatal to most species



Turbidity is the cloudiness or haziness of water caused by individual particles (suspended solids) that are generally invisible that may be invisible to the naked eye, similar to smoke in air. The measurement of turbidity is a key test of water quality. A measure of 0 to 10 jtu is considered in general terms acceptable for streams.

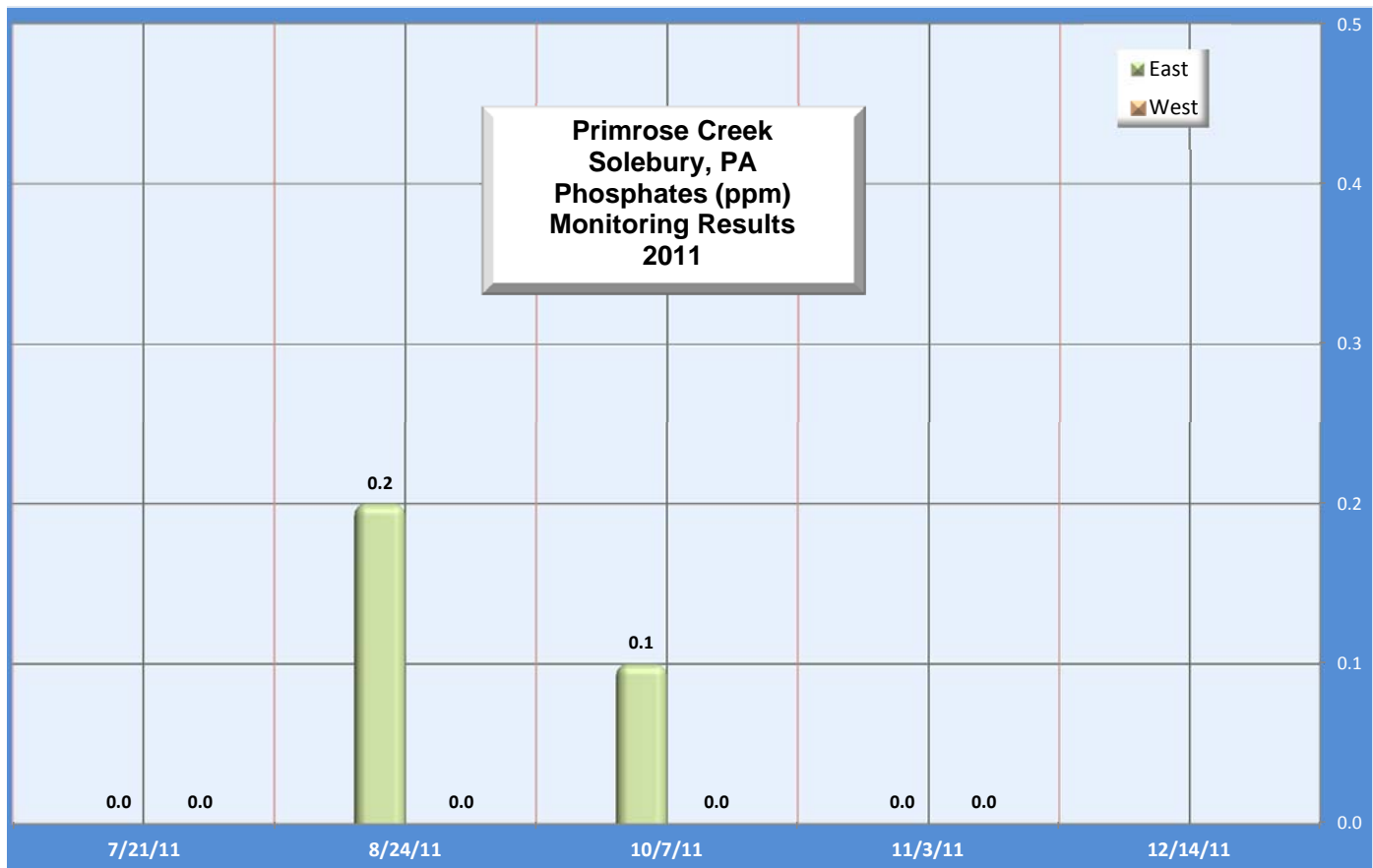


Aquatic organisms benefit from a stable pH value in their optimal range. To maintain a fairly constant pH in a water body, a higher alkalinity is preferable. High alkalinity means that the water body has the ability to neutralize acidic pollution from rainfall or basic inputs from wastewater.



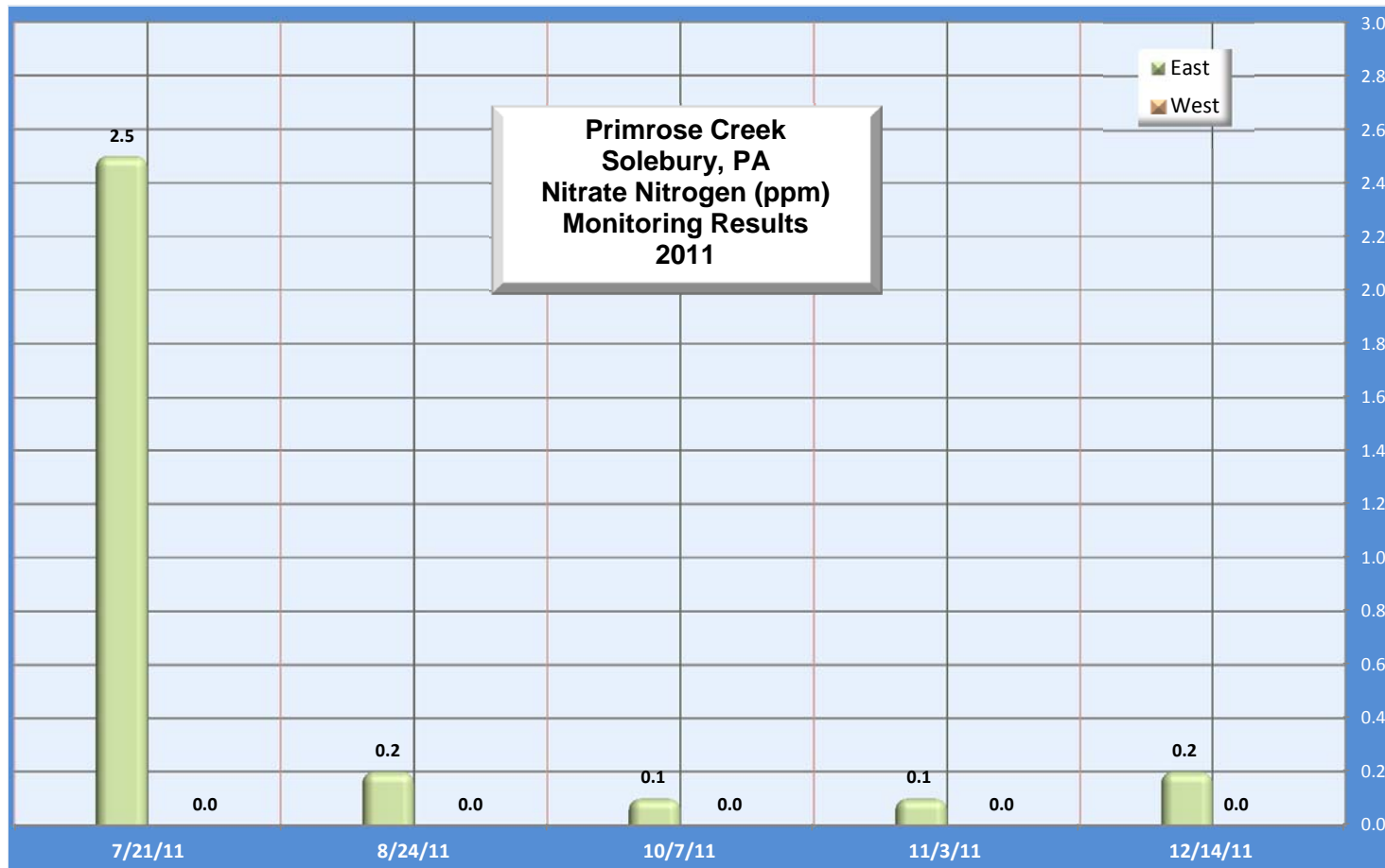
Water hardness is the indication of mineral content.  
 < 17.1 Soft  
 17.1 - 60 Slightly Hard  
 60 - 120 Moderately Hard  
 120 - 180 Hard  
 over 180 Very Hard





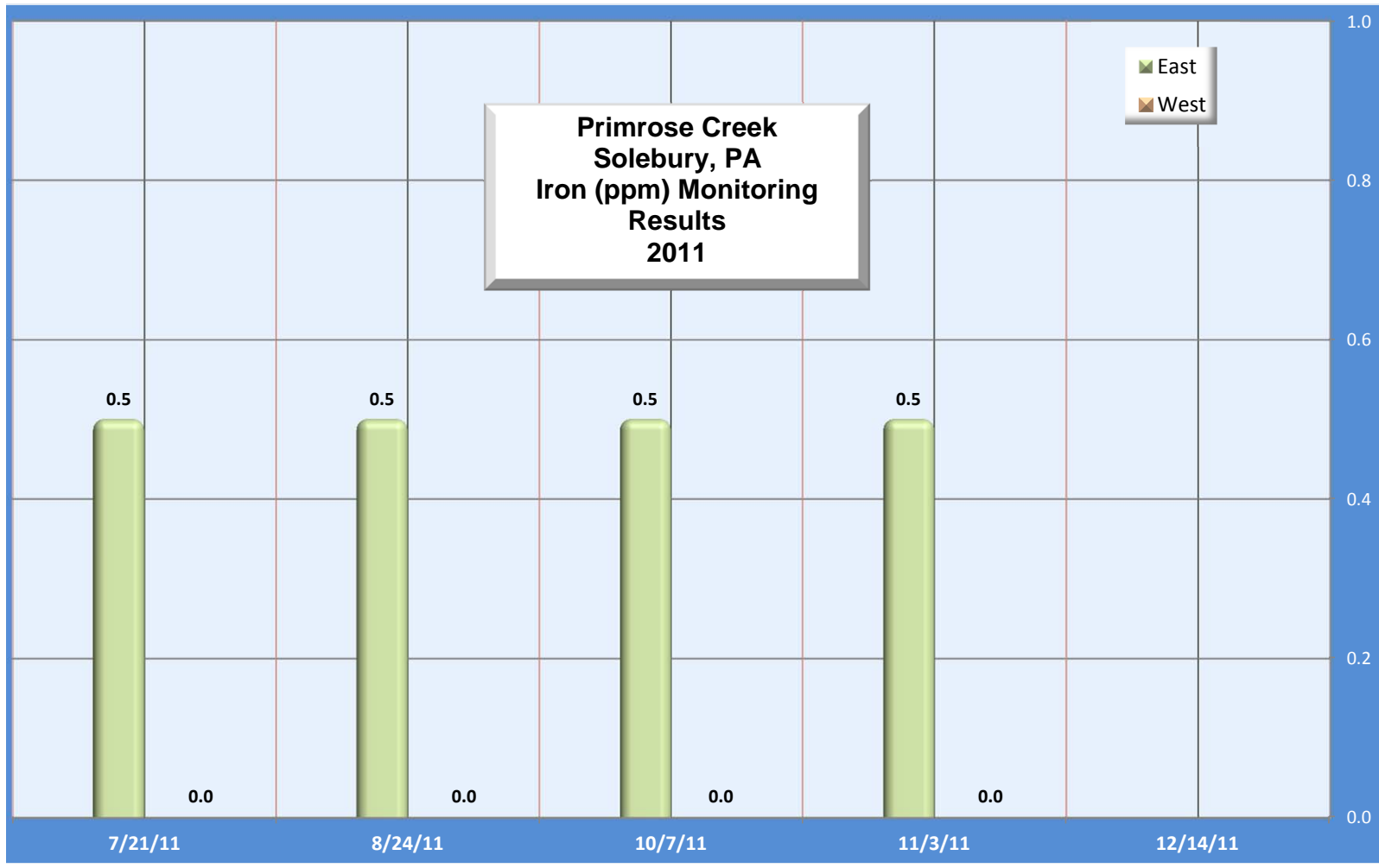
Acceptable range is 0 - .1

Phosphorus comes from both point and nonpoint sources. *Point sources* include municipal waste treatment plants, industrial discharge, large confined livestock operations, and urban stormwater. *Nonpoint sources* of phosphorus include soil erosion and water runoff from cropland, lawns and gardens, home waste treatment systems, livestock pastures, rangeland, and even forests.



Acceptable level - 0 - 10.

Sources of nitrate in water are from natural as well as human activities. Human activities include fertilizers, animal feedlots, septic systems, wastewater treatment lagoons, animal wastes and industrial wastes.



The recommended limit for iron in water, 0.3 ppm, is based on taste and appearance rather than on any detrimental health effect.

## **Source Data**

# Primrose Creek Monitoring Data Summary - 2011

East Team - David Harrison, Robert Long

West Team - Carol Cope, Charles Furst

	<u>East</u>	<u>West</u>
	<b>7/21/2011</b>	<b>7/20/2011</b>
PH (s.u.)	8.0	7.0
Disolved Oxygen (ppm)	6.2	4.2
Turbidity (jtu)	0.0	0.0
Alkalinity (ppm)	164.0	100.0
Hardness (ppm)	182.0	184.0
Phosphates (ppm)	0.0	0.0
Nitrate Nitrogen (ppm)	2.5	0.0
Iron (ppm)	0.5	0.0
Time	4:30pm	6pm
<i>Weather</i>	Hot, Humid	90 degrees
<i>Cloud Cover (%)</i>	0.0	0.0
<i>Wind Conditions</i>	light	none
<i>Rain Conditions</i>	none	none
<i>Staff Gage</i>	0.6	n/a
<i>Flow Regime</i>	Baseflow	
<i>Collector</i>	RL	CF/CBC
	<b>8/24/2011</b>	<b>8/24/2011</b>
PH (s.u.)	7.5	7.0
Disolved Oxygen (ppm)	3.2	7.6
Turbidity (jtu)	0.0	0.0
Alkalinity (ppm)	184.0	62.0
Hardness (ppm)	184.0	105.0
Phosphates (ppm)	0.2	0.0
Nitrate Nitrogen (ppm)	0.2	-
Iron (ppm)	0.5	-
Time	5:30pm	
<i>Weather</i>	Clear, 82 Degrees	Sunny, 75-80
<i>Cloud Cover (%)</i>	0.0	
<i>Wind Conditions</i>	light	light
<i>Rain Conditions</i>	none	
<i>Staff Gage</i>	0.8	
<i>Flow Regime</i>	Baseflow	
<i>Collector</i>	DH/RL	CBC

	<b>10/7/2011</b>	<b>10/7/2011</b>
PH (s.u.)	7.3	7.0
Disolved Oxygen (ppm)	12.8	9.9
Turbidity (jtu)	0.0	0.0
Alkalinity (ppm)	222.0	50.0
Hardness (ppm)	180.0	120.0
Phosphates (ppm)	0.1	0.0
Nitrate Nitrogen (ppm)	0.1	0.0
Iron (ppm)	0.5	-
Time	11am	5pm
<i>Weather</i>	Clear, 55	sunny, 70
<i>Cloud Cover (%)</i>	0.0	0.0
<i>Wind Conditions</i>	none	none
<i>Rain Conditions</i>	none	none
<i>Staff Gage</i>	1.0	-
<i>Flow Regime</i>	Baseflow + water slightly turbid	
<i>Collector</i>	RL	CBC

	<b>11/3/2011</b>	<b>11/3/2011</b>
PH (s.u.)	8.0	7.0
Disolved Oxygen (ppm)	14.0	9.2
Turbidity (jtu)	0.0	0.0
Alkalinity (ppm)	148.0	62.0
Hardness (ppm)	195.0	117.0
Phosphates (ppm)	0.0	0.0
Nitrate Nitrogen (ppm)	0.1	0.0
Iron (ppm)	0.5	-
Time	3:15pm	4:30pm
<i>Weather</i>	Partly Cloudy, 58	Partly Cloudy, approx 60
<i>Cloud Cover (%)</i>		
<i>Wind Conditions</i>	light	light
<i>Rain Conditions</i>	none	none
<i>Staff Gage</i>	0.86	
<i>Flow Regime</i>		
<i>Collector</i>	DH/RL	CF/CBC

	<b>12/14/2011</b>	<b>12/14/2011</b>
PH (s.u.)		
Disolved Oxygen (ppm)	11.8	10.0
Turbidity (jtu)	20.0	0.0
Alkalinity (ppm)	-	-
Hardness (ppm)	-	-
Phosphates (ppm)	-	-
Nitrate Nitrogen (ppm)	0.2	0.0
Iron (ppm)	-	-
Time	3:30pm	5:00pm

<i>Weather</i>	Mostly Cloudy, 50	dark
<i>Cloud Cover (%)</i>		
<i>Wind Conditions</i>	light	
<i>Rain Conditions</i>	none	
<i>Staff Gage</i>		
<i>Flow Regime</i>		
<i>Collector</i>	DH/RL	CF/CBC